

## COULOMETER FOR SULFUR MEASUREMENTS



**Figure 1:** Model CM5017S Sulfur Coulometer

The UIC, Inc. Sulfur Coulometer quantitatively titrates SO<sub>2</sub> and H<sub>2</sub>S. Typical applications include the determination of total sulfur (by combustion) and the determination of SO<sub>2</sub> and H<sub>2</sub>S (by acid evolution).

### **PRINCIPLES OF OPERATION**

The coulometer cell is filled with a solution which initially contains a slight excess of free iodine. When SO<sub>2</sub> or other reducing substances enter the cell, iodine is consumed. The amperometric-sensing circuit detects the deficiency of iodine in the solution and causes iodine to be electrically generated at a rate proportional to the sensed deficiency. When all of the substance has been titrated, the iodine is restored to its initial concentration, and the quantity of the titration is read directly on the display in user-selectable units. Since the coulometric efficiency is 100 percent, sample calibration is not necessary. The linear range and accuracy of the coulometric technique exceeds that obtained by other detection methods.

### **EASE OF OPERATION**

The amount of cell titrating solution used daily is dependent upon the amount of sulfur in the samples analyzed. It can be changed daily, or as seldom as once every few days. The cell is ruggedly constructed and refilling the cell is a fast and simple operation. The Sulfur Coulometer is factory calibrated and does not require daily field or lab calibration. All peripherals are designed for ease of operation.

### **RATE AND RANGE**

The range of the Sulfur Coulometer is from 0.01 micrograms to over 100 milligrams S. At its maximum rate (240 mA), the Coulometer can titrate 2,400 micrograms S per minute. The proportioning circuitry allows for fast and accurate completion of the titration. The S Coulometer has an accuracy of .15% +/-2 digits for standard range.

Contact us for more information:

## INTERFERENCES

Any substance that reduces iodine will be determined by the Coulometer. Interfering substances are removed by scrubbing systems. Titration is normally used to determine  $\text{SO}_2$  or  $\text{H}_2\text{S}$ , and the titration of either one involved a 2-electron transfer, giving the same titration in terms of quantity of S.

## APPLICATIONS

1. **Total Sulfur** - The Sulfur Coulometer can be used to detect  $\text{SO}_2$  from high-temperature combustion sulfur analyzers. With most high-temperature combustion units, only 70 - 90% S is converted to  $\text{SO}_2$ . These systems must be calibrated for each sample matrix used to compensate for varying  $\text{SO}_2$  conversion efficiencies. UIC, Inc.'s Sulfur Coulometer does not need to be calibrated to compensate for varying conversion efficiencies.

The most desirable systems are those which quantitatively convert all S to  $\text{SO}_2$ , taking advantage of the absolute determination provided by the Sulfur coulometer. Two approaches to this idea have been used:

- a. Combusting the sample in excess oxygen, followed by using an inert gas to pass the combustion products over metallic copper for removal of excess oxygen and quantitative conversion of S to  $\text{SO}_2$  before passing into the Coulometer.
- b. Mixing the sample with oxygen donors, such as vanadium pentoxide ( $\text{V}_2\text{O}_5$ ) and manganese dioxide ( $\text{MnO}_2$ ), and using an inert carrier gas to pass the pyrolysis products over copper and into the Sulfur Coulometer.

In both above systems, sulfur is quantitatively converted to  $\text{SO}_2$ , which is titrated by the Sulfur Coulometer, giving a direct readout. It is interesting to note that Sulfur Coulometer could be used to titrate  $\text{H}_2\text{S}$  from hydrogenation, and the titration factor is the same as for the  $\text{SO}_2$  titration.

2.  **$\text{SO}_2$**  - in solid sulfites and in sulfite solutions have been determined using the UIC, Inc. Model 5330 Acidification Module as the sample handling peripheral. The sample is acidified and an inert gas ( $\text{N}_2$ ) is used to transfer the  $\text{SO}_2$  from the reaction tube to the Coulometer. This method provides high accuracy assays of sulfite chemicals, as well as sulfite/sulfide concentrations in various industrial and natural liquids.
3.  **$\text{H}_2\text{S}$**  - in solids and solutions is analyzed similarly to sulfites using the Model 5330 Acidification Module for sample handling. Although many applications require determining total  $\text{SO}_2$  and  $\text{H}_2\text{S}$ , it is possible to separate the two gases such that only one is determined by the Sulfur Coulometer.

## ADVANTAGES

1. Factory calibration - does not require sample standardization or calibration
2. Rapid analysis time
3. User friendly
4. Minimum maintenance
5. High reliability
6. High sensitivity

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